

3-Methoxy-19-norpentara-1,3,5(10),17(20)-
tetraen-21-oneMasataka Watanabe,^a Taisuke
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Key indicators

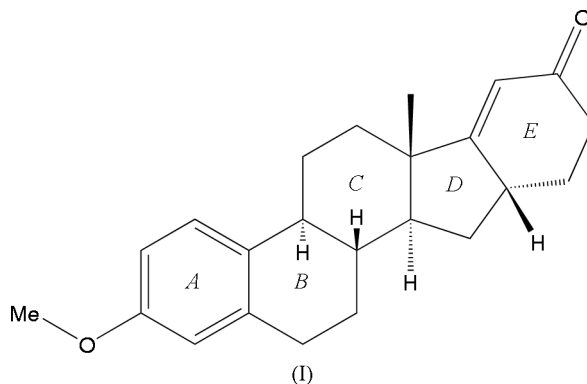
Single-crystal X-ray study
 $T = 123\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$
 R factor = 0.034
 wR factor = 0.073
Data-to-parameter ratio = 9.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the title compound, alternatively called 3'-oxocyclohex-1'-eno-1',6':16,17-3-*O*-methylestra-1,3,5(10),16(16*H*)-tetraen-3-ol, $\text{C}_{23}\text{H}_{28}\text{O}_2$, the cyclohexenone ring is fused to the five-membered ring through the α position at C16. As a result of the ring annelation, the distance between carbonyl and methoxy O atoms [12.994 (2) \AA] is much longer than that in estrones or estradiol. The crystal packing is stabilized by $\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ hydrogen-bonding interactions.

Received 3 December 2004
Accepted 17 December 2004
Online 24 December 2004

Comment

The title compound, (I), was synthesized by Allison *et al.* (1967) as a potential anti-inflammatory agent. At the time, the stereochemistry of the introduced chirality at C16 was determined through ORD spectroscopic analysis of the $n-\pi^*$ transition of the C17(20)-en-21-one system. An X-ray crystal structure analysis of (I) was carried out in order to better understand interactions of (I) with steroidal receptors.



Ring A (Fig. 1) is essentially planar, as observed in other estrones and estradiols. Ring B has a half-chair conformation, with Cremer & Pople (1975) puckering parameters $Q = 0.507(3)\text{ \AA}$, $\theta = 47.5(3)^\circ$ and $\varphi = 154.1(4)^\circ$. This conformation

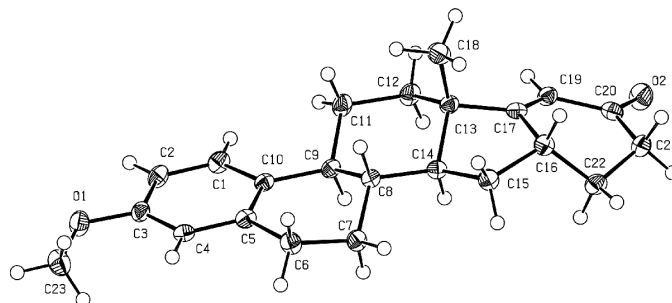


Figure 1
View of (I) (Spek, 2003), showing 50% probability displacement ellipsoids and the atomic numbering.

is also in accordance with the relative signs of the endocyclic torsion angles within ring *B* (Boeyens, 1978). Ring *C* adopts the chair conformation [$Q = 0.588$ (3) Å, $\theta = 4.2$ (3)° and $\varphi = 311$ (3)°]. Ring *D* is in a half-chair conformation [$Q = 0.451$ (3) Å and $\varphi = 191.0$ (3)°], with a pseudorotation angle $\Delta = 352.0$ (2)°, and a maximum torsion angle $\varphi = 46.4$ (1)° (Rao *et al.*, 1981) for the atom sequence C13–C14–C15–C16–C17.

Ring *E* adopts an envelope conformation with puckering parameters $Q = 0.474$ (3) Å, $\theta = 55.7$ (3)° and $\varphi = 301.8$ (4)°. Ring *E* is connected to ring *D* through the α position at C16. This stereochemistry at C16 had been determined by Allison *et al.* (1967) correctly through the ORD measurements (see above). Atoms C17, C19, C20 and O2 are coplanar [C17–C19–C20–O2 = -178.2 (2)°] due to the π conjugation of the enone moiety. Atom C22 is out of plane, minimizing the ring strain to give the envelope conformation of ring *E*.

Typically, the oxygen at C3 (O1) and the oxygen on the ring furthest from ring *A* (oxygen at C17 in estrone and estradiols) significantly influence the binding of the steroid to the steroidal receptor. In common estra-1,3,5(10)-trien-17-one (estrone) derivatives, the direction of the C=O bond is almost parallel to the plane determined by ring *A* or slightly directed towards the α -side of that plane [C3–C2–C17–O2 torsion angles range from -168.3 to -178.9 °, with an average of -174 ° (van den Bossche, 1971; Busetta *et al.*, 1973; Debaerdemaeker, 1972)]. In (I), the corresponding torsion angle C3–C2–C20–O2 is -159.4 (2)°, which is much larger than in usual estrones (see above), but less than in other estra-3,17 α -diol derivatives (124.2 and 117.3° for C3–C2–C17–O2; Busetta *et al.*, 1976).

As a result of the *E*-ring annelation, the O1···O2 distance of 12.994 (2) Å in (I) is much longer than that in estrones [10.79–10.90 Å, average 10.83 Å; van den Bossche, 1971; Busetta *et al.*, 1973; Debaerdemaeker, 1972], estra-3,17 α -diols (10.54 and 10.32 Å; Busetta *et al.*, 1976) and estra-3,17 β -diols (10.83–11.06 Å, average 10.99 Å; Bolaños-García *et al.*, 1996; Busetta, Courseille *et al.*, 1972; Busetta & Hospital, 1972; Duax, 1972; Parrish & Pinkerton, 1999; Parrish, 2003; Prokai *et al.*, 2001; Punzi *et al.*, 1992; Reck *et al.*, 1986; Starova *et al.*, 2001; Tsukuda *et al.*, 1968; Vichard *et al.*, 1992; Zacharias *et al.*, 1995).

In the crystal structure, molecules translated by a unit along the *c* axis are linked by C4–H4···O2ⁱ hydrogen bonds to form a chain. Adjacent chains are linked *via* C9–H9···O2ⁱⁱ hydrogen bonds and C22–H22A···Cg1ⁱⁱⁱ (Cg1 is the C1–C6 ring centroid) interactions to form a layered structure. Symmetry codes and the geometry for these interactions are given in Table 1.

Experimental

Compound (I) was prepared by the Robinson annelation procedure from 16-hydroxymethylidene-3-methoxyestra-1,3,5(10)-trien-17-one by a known procedure (Allison *et al.*, 1967). The crystal used for X-ray structure analysis was obtained by recrystallization of (I) from chloroform–ether–hexane (1:1:1).

Crystal data

C₂₃H₂₈O₂
M_r = 336.47
 Orthorhombic, *P*2₁2₁2₁
a = 10.980 (4) Å
b = 12.499 (4) Å
c = 12.865 (4) Å
V = 1765.5 (10) Å³
Z = 4
D_x = 1.266 Mg m⁻³

Mo *K*α radiation
 Cell parameters from 5425 reflections
 $\theta = 3.2$ – 27.5 °
 $\mu = 0.08$ mm⁻¹
T = 123.1 K
 Prism, colourless
 0.16 × 0.12 × 0.05 mm

Data collection

Rigaku Saturn diffractometer
 ω scans
 Absorption correction: multi-scan (Jacobson, 1998)
T_{min} = 0.879, *T_{max}* = 0.996
 14467 measured reflections
 2295 independent reflections

1431 reflections with $F^2 > 2\sigma(F^2)$
R_{int} = 0.053
 $\theta_{\text{max}} = 27.5$ °
h = $-14 \rightarrow 14$
k = $-16 \rightarrow 16$
l = $-13 \rightarrow 16$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.073$
S = 1.00
 2295 reflections
 255 parameters
 All H-atom parameters constrained

$w = 1/[0.3590\sigma(F_o^2)]/(4F_o^2)$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.26$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³
 Extinction correction: Larson (1970), equation 22
 Extinction coefficient: 54.9 (11)

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C4–H4···O2 ⁱ	0.93	2.54	3.410 (3)	155
C9–H9···O2 ⁱⁱ	0.98	2.69	3.627 (3)	160
C22–H22A···Cg1 ⁱⁱⁱ	0.97	2.52	3.414 (3)	152

Symmetry codes: (i) $x, y, z - 1$; (ii) $\frac{1}{2} - x, -y, z - \frac{1}{2}$; (iii) $\frac{1}{2} - x, -y, \frac{1}{2} + z$. Cg1 is the C1–C6 ring centroid.

H atoms were placed in calculated positions (C–H = 0.93–0.98 Å) and were included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H})$ values set at 1.2 $U_{\text{eq}}(\text{carrier atom})$. In the absence of significant anomalous dispersion effects, Friedel pairs were averaged. The absolute configuration was assigned from that of the starting material in the synthesis.

Data collection: *CrystalClear* (Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MSO, 2004); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *CrystalStructure*.

References

- Allison, J. M., Burn, D., Butcher, F. K., Davies, M. T. & Petrow, U. (1967). *Tetrahedron*, **23**, 1515–1534.
 Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
 Boeyens, J. C. A. (1978). *J. Cryst. Mol. Struct.* **8**, 317–320.
 Bolaños-García, V. M., Juárez-Martínez, G., Panneerselvam, K. & Soriano-García, M. (1996). *Acta Cryst.* **C52**, 1997–2000.
 Bossche, G. van den (1971). *Bull. Soc. R. Sci. Liege*, **40**, 614–627.
 Busetta, B., Barrans, Y., Precigoux, G. & Hospital, M. (1976). *Acta Cryst.* **B32**, 1290–1292.

- Busetta, B., Courseille, C., Geoffre, G. & Hospital, M. (1972). *Acta Cryst.* **B28**, 1349–1351.
- Busetta, B., Courseille, C. & Hospital, M. (1973). *Acta Cryst.* **B29**, 298–313.
- Busetta, B. & Hospital, M. (1972). *Acta Cryst.* **B28**, 560–562.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Debaerdemaeker, T. D. J. (1972). *Cryst. Struct. Commun.* **1**, 39–42.
- Duax, W. L. (1972). *Acta Cryst.* **B28**, 1864–1871.
- Jacobson, R. (1998). Private communication to the Rigaku Corporation.
- Larson, A. C. (1970). *Crystallographic Computing*, edited by F. R. Ahmed, S. R. Hall & C. P. Huber, pp. 291–294. Copenhagen: Munksgaard.
- Parrish, D. A. & Pinkerton, A. A. (1999). *Acta Cryst.* **C55**, IUC990010.
- Parrish, D. A. & Pinkerton, A. A. (2003). *Acta Cryst.* **C59**, o80–o82.
- Prokai, L., S. Oon, S.-M., Prokai-Tatrai, K., Abboud, K. A. & Simpkins, J. W. (2001). *J. Med. Chem.* **44**, 110–114.
- Punzi, J. S., Duax, W. L., Strong, P., Griffin, J. F., Flocco, M. M., Zacharias, D. E., Carrell, H. L., Tew, K. D. & Glusker, J. P. (1992). *Mol. Pharmacol.* **41**, 569–576.
- Rao, S. T., Westhof, E. & Sundaralingam, M. (1981). *Acta Cryst.* **A37**, 421–425.
- Reck, G., Schubert, G. & Bannier, G. (1986). *Cryst. Res. Technol.* **21**, 1313–1319.
- Rigaku (1999). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSK (2004). *CrystalStructure*. Version 3.6.0. Rigaku/MSK, 9009 New Trails Drive, The Woodlands, TX 77381–5209, USA.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Starova, G. L., Eliseev, I. I., Abusalimov, Sh. N., Tsogoeva, S. B. & Shavva, A. G. (2001). *Kristallografiya*, **46**, 72–75.
- Tsukuda, Y., Sato, T., Shiro, M. & Koyama, H. (1968). *J. Chem. Soc. B*, pp. 1387–1393.
- Vichard, D., Gruselle, M., El Amouri, H., Jaouen, G. & Vaissermann, J. (1992). *Organometallics*, **11**, 976–979.
- Zacharias, D. E., Glusker, J. P., Tew, K. T. & Hartley-Asp, B. (1995). *Struct. Chem.* **6**, 371–376.